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? ? (5n ? 6) 8531) n = 12 2) 12 3) 4) n ? 1 16656) a. 2.048 ft b. 131.968 ft c. 2.5 min d. 133.333 ft) 81x4
 216x3y + 216x2y2 96xy3 + 16y4 8) -960a3b79) 10P3 = 720 10) 413C5 = 514811) a. 5C5 26C5 = 1 65780 b.
 (5C321C2+5C421C1+5C5) 26C5 = 0.0335c. 1 P (no vowels) = 1 (21C5 26C5) = 0.69112) a. 30C10 b. 12C5
 18C5 c. 18C312C1 + 18C413) a. 13 52 + 4 52 1 52 = 4 13 b. (13C2+4C2) 52C2 = 14 221 = 0.06314...

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[UNIT 4 PRACTICE TEST](#)

Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear Chemistry Part A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response) ?Ebind = ??mc2
 c = speed of light (3.00 108 m s) 1 eV = 1.69 10?19 J (1 kg = 6.022 1026 amu , or 1 g = 6.022 1023 amu = 1 mole amu) 1 MeV = 1.69 10?16 J (1 Neutron = 1.008665 amu) (1 Proton = 1.007825 am...

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Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases 1 atm = 760 mm Hg = 760 torr = 101.325 kPa TK = TC + 273.15 STP = 22.4 L mol @ 0.00C & 101.325 kPa (1 atm) SATP = 24.8 L mol @ 25.0C & 100. kPa L atm L kPa m m nsolute R = 0.0821 or 8.314 D= n= C= K mol K mol V M Vsolution P1V1 P2V2 r1 M2 n1 P1 C1V1 = C2V2 PV = nRT = = ?1 = = n1T1 n2T2 r2 M1 nTotal PTotal PTotal = P1 +...





















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