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UNIT 1 EXAM PREPARATION INSTRUCTIONS

Unit 1 Exam - Thursday, September 8 Review Sessions: Wednesday, September 7, AM PM room A 211, Tuesday, September 6, 3:30 PM EL lab This exam is closed notes book handouts, except where noted Structure of the Exam 1. 25 multiple choice matching questions drawn from the terms and questions from the unit outline as indicated in class. You will see these questions on the day before the exam. 2. You will...

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UNIT 1: BASIC CHEMISTRY NOTES (ANSWERS)

Honour Chemistry Unit 1: Basic Chemistry Unit 1: BASIC CHEMISTRY Chapter 3: Scientific Measurement 3.1: The Importance of Measurement Qualitative Measurements: - measurements that do not involve a numerical value. (Examples: Colour, Odour, Heat Given off or Taken in, Type of Solid Formed etc) Quantitative Measurements: - measurements that do include a numerical value. (Examples: Volume, Temperature, M...

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UNIT 4 CHEMICAL BONDING NOTES (ANSWERS)

Unit 4: Chemical Bonding Honour Chemistry UNIT 4: CHEMICAL BONDING Chapter 13: Electrons in Atoms 13.1: Models of Atom As mentioned in Section 5.2: Structure of the Nuclear Atom, the evolution on the Atomic Theories is as follows: 1. Daltons Atomic Model: - John Dalton theorized that atoms are the smallest fundamental particle unique to different elements. They can never be created or destroyed. 2. Thom...

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UNIT 5: ORAGNIC CHEMISTRY NOTES (ANSWERS)

Honour Chemistry Unit 5: Organic Chemistry UNIT 5: ORGANIC CHEMISTRY Chapter 25: Hydrocarbon Compounds 25.1: Hydrocarbons Hydrocarbons: - compounds that contains hydrogen and carbon atoms. - it may contain oxygen, nitrogen and other halogen atoms. In complex organic compound, it may even contain transition metals. Examples: CH₄ (Methane), C₃H₈ (Propane), C₆H₁₂O₆ (Glucose), CH₃OH (Methanol) are hydroca...





















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